

ARTICLES

Anthocyanins from Bambara Groundnut (*Vigna subterranea*)Eloi Pale,[†] Mouhoussine Nacro,[†] Maurice Vanhaelen,^{*,‡} and Renée Vanhaelen-Fastré[‡]

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Among five pigments detected in bambara groundnut (*Vigna subterranea*), three major anthocyanins were isolated by combined column chromatography on Amberlite XAD-7, Sephadex LH-20, and RP-18 silica gel and by preparative thin-layer chromatography on silica gel. They were identified as delphinidin 3-*O*- β -glucoside, petunidin 3-*O*- β -glucoside, and malvidin 3-*O*- β -glucoside on the basis of chromatographic data and UV-vis, ¹H-NMR, and mass spectrometry.

Keywords: *Vigna subterranea*; Fabaceae; delphinidin; petunidin; malvidin; glucoside; ¹H NMR

INTRODUCTION

The grains of *Vigna subterranea* [L.] Verdc. (Fabaceae) (Mabberley, 1987) are widely used as food in West Africa. Previous works on other *Vigna* species have led to the isolation of cyanidin, malvidin, and delphinidin glycosides (Mazza and Miniati, 1996). Delphinidin 3-glucoside was found in *V. mungo*, *V. radiata*, and *V. angularis*; cyanidin 3-glucoside was identified in *V. mungo* and *V. radiata*; delphinidin 3-glucoside was isolated from *V. angularis* and *V. radiata*; delphinidin 3-(*p*-coumaroyl)glucoside was detected in *V. mungo* and *V. radiata*, while delphinidin 3,5-diglucoside was found only in *V. radiata*.

To complete the chemical investigation of the genus *Vigna* and to pursue our studies on the pigment content of several plants growing in Burkina Faso, we report here the identification of 3- β -*O*-monoglucosides of cyanidin, petunidin, and malvidin in *V. subterranea* grains.

MATERIALS AND METHODS

Reagents and Solvents. All were of analytical grade and obtained from Merck-Clevenot Corp., Darmstadt, Germany.

Sample Preparation. Extracts were obtained from grains of *V. subterranea* collected in 1995 in Burkina Faso, West Africa. The pigments were extracted twice by maceration of 600 g of the grains with 1% trifluoroacetic acid (TFA) in CH₃-OH at 5 °C for 48 h. The crude extract was concentrated to dryness under vacuum and dissolved in 10 mL of 0.5% TFA in H₂O. This solution was filtered and fixed on a nonionic polymeric adsorbent (Amberlite XAD-7, Aldrich Chemical Co., Milwaukee, WI) column (length 300 mm, i.d. 20 mm) which was prewashed with 0.5% TFA/H₂O; the pigments were then eluted with MeOH/H₂O/TFA, 70:30:0.3). The eluate was concentrated, filtered through a Sephadex LH-20 (Pharmacia Biotech, Uppsala, Sweden) column (length 300 mm, i.d. 20 mm), and further purified on a Lobar glass prepac column (Licroprep RP-8 column, 310 mm length, 25 mm i.d., Merck-

Clevenot Corp.) with MeOH/H₂O/TFA, 40:60:0.3. The final purification was achieved by preparative thin-layer chromatography (TLC) on silica gel 60 F₂₅₄ (Merck-Clevenot Corp.) using EtOAc/HCO₂H/HOAc/H₂O, 100:11:101:26 (EFAW), as solvent system. The isolated bands of adsorbent were eluted with 0.5% TFA/MeOH, and the solutions were concentrated and filtered through a RP-18 column using (0.5% TFA/H₂O)/MeOH (6:4). The eluates were finally concentrated and freeze-dried to give **1** (5 mg), **3** (10 mg), and **5** (4 mg) as TFA salts.

HPLC and GLC Conditions. High-performance liquid chromatography (HPLC) was carried out following the method described by Gao and Mazza (1994); the elution conditions were, however, modified according to the polarity of the pigments under investigation. The chromatographic system was equipped with two pumps (Model 6000 A), a solvent programmer (Model 660) from Waters Associates (Milford, MA), a diode array detector operating at 520 nm (Model 1040A from Hewlett-Packard, Germany), and a Nova Pak C₁₈ column (150 × 3.9 mm; mean particle size = 4 μ m); the elution was performed at 25 °C at a flow rate of 0.8 mL min⁻¹. A linear gradient solvent system was applied for 20 min from 20 to 60% of solvent B (MeOH/H₂O/HCO₂H, 75:24.5:0.5) in solvent A (H₂O/HCO₂H, 60:1); the final isocratic conditions for elution were maintained for 10 min. A photodiode array detector was used for the detection (520 nm) and the UV-vis spectra recording (from 250 to 600 nm).

Gas-liquid chromatography (GLC) identification of the sugar moieties was carried out by dissolving 2 mg of each anthocyanin in 4 mL of 2 N TFA/H₂O and heating in a sealed vial at 110 °C for 45 min. The anthocyanidins were extracted twice by 0.5 mL of 3-methyl-2-butanol (Ribéreau-Gayon, 1968). The resulting aqueous fractions were evaporated to dryness. The residue was taken for silylation in pyridine/*N*-(trimethylsilyl)imidazole (2:1) and heated in a sealed vial at 60 °C for 15 min. The silylated derivatives of sugar residues were identified by GLC on a fused silica capillary column coated with CP-Sil-8-CB (length 25 m, i.d. 0.32 mm, film thickness 0.25 μ m, Chrompack, Belgium) using cochromatography with sugar standards.

Spectroscopic Analysis. ¹H NMR spectral data were recorded on an instrument operating at 250 MHz (Model WM 250, Bruker, Karlsruhe, Germany) in CD₃OD/TFA-*d*₁ (0.5:0.1 mL) with TMS as internal standard. When the signals of the anomeric protons of the investigated anthocyanins were partially overlapped by the signal of the dissolved H₂O, an additional shift to lower magnetic field of this signal was produced by adding 30 μ L of TFA-*d*₁.

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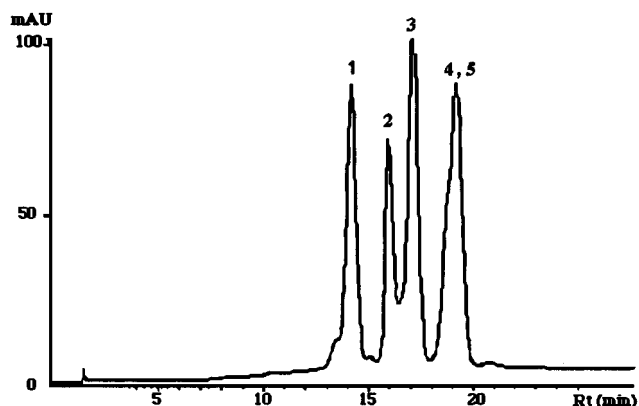
Table 1. Chromatographic and UV-Vis Absorption Data of Anthocyanins 1, 3, and 5^a

anthocyanin	TLC ($R_f \times 100$)					HPLC (R_t in min)	UV-vis data (in 0.01 N HCl-MeOH)			
	BAW ^a	BuHCl ^a	1% HCl ^a	AHW ^a	EAFW ^b		UV λ , nm (log ϵ)	vis λ , nm (log ϵ)	+AlCl ₃ ^d (Δ in nm)	$E_{440}/E_{vis,max}$
1	19	06	05	10	34	14.40	279 (4.14)	541 (4.47)	+14	16
3	25	07	05	14	39	17.10	278 (4.22)	540 (4.55)	+9	16
5	32	11	08	23	44	19.10	278 (4.14)	539 (4.43)		18

^a Solvent systems: BAW, 1-butanol/acetic acid/water (4:1:5 upper phase); BuHCl, 1-butanol/2 N HCl (1:1 upper phase); 1% HCl, water/concentrated HCl (99:1); AHW, acetic acid/concentrated HCl/water (15:3:82); EAFW, ethyl acetate/acetic acid/formic acid/water (100:11:11:26). ^b Adsorbent: cellulose microcrystalline F. ^c Adsorbent: silica gel 60. ^d 2–3 drops of 5% AlCl₃ in MeOH were added.

Table 2. ¹H-NMR Spectral Data of Anthocyanins 1, 3, and 5 [δ in CD₃OD/TFA-*d*₁ (5:1)]

anthocyanin	1 δ (J in Hz)	3 δ (J in Hz)	5 δ (J in Hz)
aglycon	delphinidin	petunidin	malvidin
H-4	8.98 s	8.89 s	8.94 s
H-6	6.68 d (1.9)	6.57 d (1.9)	6.60 d (1.9)
H-8	6.88 d (1.3)	6.78 d (1.3)	6.83 d (1.3)
H-2'	7.78 s	7.82 d (2.1)	7.87 s
H-6'	7.78 s	7.69 d (2.1)	7.87 s
O-CH ₃		3.89 s	3.91 s
3- <i>O</i> - β -D-glucopyranose			
H-1''	5.32 d (7.6)	5.23 d (7.6)	5.25 d (7.6)
H-2'', H-3'', H-4'', and H-5''	3.51–3.81	3.26–3.68	3.30–3.71
H-6a''	3.96 dd (1.3, 10.4)	3.84 dd (1.2, 12.0)	3.84 dd (2.0, 12.0)
H-6b''	3.78 d (10.4)	3.66 d (12.0)	3.65 d (12.0)

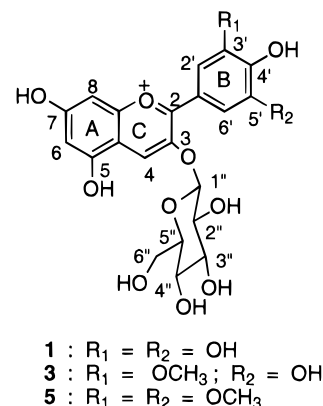
**Figure 1.** HPLC chromatogram of a crude methanol extract of grains of *V. subterranea* (conditions are given under Materials and Methods).

Liquid secondary ion mass spectrometry (SIMS) was carried out using a mass spectrometer (Model Autospec, Micromass, Manchester, England). The matrix was glycerol/HCl (1:1).

RESULTS AND DISCUSSION

V. subterranea grains were extracted by methanol with added TFA. A preliminary HPLC of the crude extract allowed the detection of five pigments (Figure 1) having UV-vis spectra typical of the anthocyanins. The crude methanol extract was chromatographed successively on an ion-exchange resin, on a Sephadex column, and on a reversed phase silica gel column to give three major pigments: **1** (relative proportion = 26%), **3** (relative proportion = 23%), and **5** (relative proportion = 36%). They were finally purified by preparative TLC on silica gel. R_f data observed in standard TLC conditions used for this group of pigments, retention times in HPLC, and absorption maxima in UV-vis spectra of **1**, **3**, and **5** are presented in Table 1.

Our attempts to isolate the minor pigments **2** (relative proportion = 8%) and **4** (relative proportion = 4%) were unsuccessful, and their identifications were therefore dropped from of the study. The complete separation of **4** from **5** was only possible by preparative TLC on silica



gel. All isolated pigments were present in the crude methanol extract as confirmed by comparison of the HPLC profiles of a crude methanol extract prepared without TFA; a possible hydrolysis during the extraction steps in the presence of TFA was thus discarded.

The hydrolysis of **1**, **3**, and **5** in TFA solution afforded glucose as confirmed by GLC analysis after silylation. The SIMS of anthocyanins **1**, **3**, and **5** showed strong $[M]^+$ ions at, respectively, m/z 465, 479, and 493, consistent with $C_{21}H_{21}O_{12}$, $C_{22}H_{23}O_{12}$, and $C_{23}H_{25}O_{12}$ molecular formulas. In addition, ions observed at m/z 303 for **1**, at m/z 317 for **3**, and at m/z 331 for **5** were assigned to a loss of $[C_6H_{10}O_5]^+$, the glucosyl moiety, in accordance with the GLC results. The linkage of the glucosyl moiety at C-3 was deduced from the chemical shift of the anomeric protons (Table 2) and by the $E_{440}/E_{vis,max}$ values (Table 1) (Ribéreau-Gayon, 1968). The β configuration of the glucose moiety of all three anthocyanins was confirmed from the magnitude ($J = 7.6$ Hz) of the $J_{1'',2''}$ coupling constant in the ¹H NMR spectra (Table 2). The ¹H NMR data related to the aglycon protons were in close agreement with those reported for other delphinidin, petunidin, and malvidin 3-*O*- β -glycosides (Kim et al., 1989; Terahara et al., 1990; Tsuda et al., 1994). The complete data, which were partially reported until now, are listed in Table 2.

Compounds **1**, **3**, and **5** were, therefore, respectively, delphinidin 3-*O*- β -glucoside, petunidin 3-*O*- β -glucoside,

and malvidin 3-*O*- β -glucoside. The coloration of different grain samples showed marked variations; therefore, the pigment concentration was assumed to be dependent on the cultivation region and on the harvest time. The usual consumption of the *V. subterranea* grains in Burkina Faso provided a mean daily intake of 5–10 mg of the pigments. Because of the known biological activities of anthocyanins (Morazzoni and Bombardelli, 1996), health benefits due to significant consumption of this food have to be considered.

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- CAS Registry No. (Supplied by the Author):** delphinidin 3-*O*- β -glucoside, 6906-38-03; malvidin 3-*O*- β -glucoside, 7228-78-6; petunidin 3-*O*- β -glucoside, 6988-81-4.

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